PHOTOCRINAMINE: A PHOTOTRANSFORMATION PRODUCT OF AN AMARYLLIDACEAE ALKALOID, CRINAMINE

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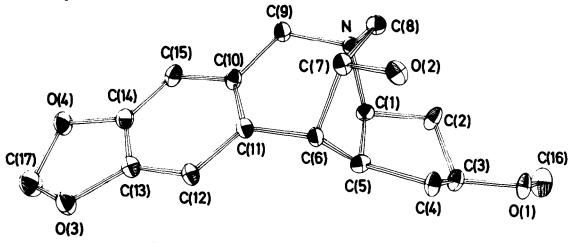
Contrary to the alkaloids of lycorine group which are considerably photosensitive, those of crinine group are believed to be stable to light¹ and none of the phototransformation product of this group has been hitherto characterized. Here we present the first example of phototransformation of an <u>Amaryllidaceae</u> alkaloid, crinamine, one of the representative alkaloid of crinine group.¹

Irradiation of crinamine (1) in methanol under N₂ with high pressure mercury lamp (200W, with Pyrex filter) for 1.5 hr yielded a crystalline product, photocrinamine (2, m.p. 189-190°) of empirical formula $C_{17}H_{21}NO_4$ (M⁺ 303) in ca. 10% yield with 20% recovery of crinamine; UV λ_{max} (MeOH) 235 sh. (loge 3.54), 295 nm (3.66); ¹H NMR(CDCl₃) δ 3.37(s, 3H, OCH₃), 4.17(s, 2H, ArCH₂N), 5.88(s, 2H, CH₂ $_{-0}^{O}$), 6.50(s, 1H, ArH), 6.56 ppm(s, 1H, ArH); ¹³C NMR(CDCl₃) δ (ppm) 146.3s, 146.0s, 135.4s, 132.2s, 108.6d, 107.1d($_{0}^{O}$), 100.8t(CH₂ $_{-0}^{O}$), 81.1d, 69.1d (2 × CH-O), 61.9t, 56.8t-q(2 × N-CH₂-), 57.9d(N-CH<), 51.3d(Ar-CH<), 49.3q (OCH₃), 39.4d(-CH<), 37.1t, 36.5t(2 × CH₂). It formed the monoacetate (3), m.p. 152°; ¹H NMR(CDCl₃) δ 5.03 ppm (m, 1H, >CH-OAc).

The spectral data of 2 and 3 indicated that the change had taken place at rings B and C as suggested from disappearance of both the double bond and the quaternary carbon in crinamine.

These results suggested the structure 2 for photocrinamine, which was confirmed by a single crystal X-ray analysis.

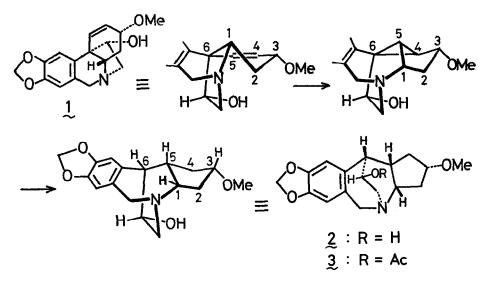
The crystals of photocrinamine (2) are orthorhombic, space group $\underline{P}_{21}^{2}_{121}^{2}_{1$



 \underline{R} to 0.086 for 1451 observed reflections. The structure thus resulted is shown in the Figure.

Figure. ORTEP³ drawing of the molecular structure of photocrinamine (2).

We consider that the reaction proceeded in $[\pi 2a + \sigma 2a]$ manner followed by photo-reduction of the intermediate aryl-conjugated cyclopropane, though there is no example of the latter reaction.



REFERENCES AND NOTES

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